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Crystallization and preliminary X-ray analysis of the thermostable sweet protein mabinlin II

Mabinlin II is a sweet protein with the highest known thermostablility and is isolated from the seeds of *Capparis masaikai* Levl. grown in south China. Two crystal forms of mabinlin II were obtained using the hanging-drop vapour-diffusion method. One of them diffracts to 2.8 Å resolution and belongs to space group P2, with unit-cell parameters a = 50.16, b = 50.17, c = 76.60 Å, $\beta = 99.6^{\circ}$. There are four molecules per asymmetric unit, with a solvent content of 35.3%.

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1. Introduction

The mature seeds of mabinlang (C. masaikai Levl.) have a unique taste. Immediately after chewing, a bitter and astringent stimulus is left on the tongue followed by a delightful and prolonged sensation of sweetness. The mature seeds are also used as a traditional Chinese medicine. Hu and coworkers (Hu & He, 1983; Hu et al., 1985) isolated two sweet proteins, named mabinlin I and II (Mab I and II), from these seeds and demonstrated that Mab II was highly heat stable with a strong sweetness property (Ding & Hu, 1986). The sweetness of Mab II was retained in solution at 353 K for at least 48 h, but Mab I lost its sweetness rapidly at 353 K. The complete amino-acid sequence determination (Liu et al., 1993; Liu & Hu, 1988; Nirasawa et al., 1994), the cDNA cloning and sequencing (Nirasawa et al., 1996) and chemical synthesis (Kohmura & Arivoshi, 1998) of Mab II have been carried out. We now know that Mab II is composed of an A chain of 33 amino-acid residues and a B chain of 72

amino-acid residues, with a total molecular weight of 12.4 kDa (Liu et al., 1993). The B chain contains two intrachain disulfide bonds and connects with the A chain through two interchain disulfide bridges (Nirasawa et al., 1993). Obviously, the next important step towards understanding the structure–function relationship of this interesting protein is to determine its three-dimensional structure. As a first step towards this, we report here the crystallization and preliminary X-ray analysis of the thermostable sweet protein mabinlin II.

Until now, six proteins, including mabinlin II, have been identified as sweet proteins: thaumatin (Van der Wel & Loeve, 1972), monellin (Morris & Cagan, 1972), mabinlin (Hu et al., 1985; Liu et al., 1993), curculin (Yamashita et al., 1990), pentadin (Van der Wel et al., 1989) and brazzein (Ming & Hellekant, 1994). Interestingly, they exhibit no significant similarities in aminoacid sequence. To date, only two crystal structures, thaumatin (Ogata et al., 1992) and monellin (Somoza et al., 1993), and one NMR structure, brazzein (Caldwell et al., 1998), have been determined. There is also no obvious structural resemblance between the proteins. Thus, the molecular mechanism and structural basis of the sweet taste remain unknown. Therefore, more detailed structural information on sweet proteins is needed. Furthermore, sweet proteins are of great value as low-calorie sweeteners. Several of them have been produced by gene expression (Kondo et al., 1997; Hahm & Batt, 1990) or by solid-phase chemical synthesis (Kohmura & Ariyoshi, 1998; Izawa et al., 1996; Kohmura et al., 1991). Determining the three-dimensional structures of more sweet proteins may provide clues about the sweetness determinant and may also

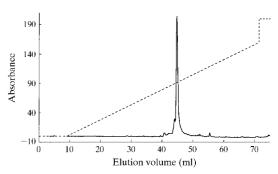


Figure 1 Chromatography of mabinlin II on an analytical reverse-phase C8 column (5 μm , 4.6 \times 250 mm) on an ÅKTA Purifier system showing the high purity of the sample used in crystallization. The solvents were 0.1% TFA (buffer A) and 80% acetonitrile containing 0.1% TFA (buffer B). The protein was eluted with a linear gradient of buffer B (shown by the dashed line) at a flow rate of 1.0 ml min $^{-1}$.

© 2000 International Union of Crystallography Printed in Denmark – all rights reserved lead to the production of new low molecularweight sweeteners.

2. Experimental and results

2.1. Purification

The mabinlin II used in crystallization was purified from the mature seeds of C. masaikai plants grown in the south of Yunnan in China according to the method described previously (Hu et al., 1985). The seed powder, defatted with petroleum ether (boiling point 303-333 K), was extracted with 50% aqueous acetone at room temperature and then precipitated at pH 10.0. After the sediment had been dissolved at pH 6.0, the sample was applied to a carboxymethylcellulose CM-52 (Whatman Inc.) column eluted with a stepwise gradient of NaCl. Mabinlin II eluted out at 0.45 M NaCl. After desalting, the sample used in the crystallization was obtained. When analyzed on a C8 reverse-phase column on an AKTA

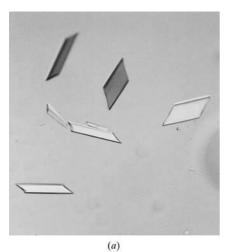




Figure 2 Crystal photographs of mabinlin II. (*a*) Crystal form I, (*b*) crystal form II.

Purifier system (Amersham Pharmacia Biotech), the protein showed a high degree of purity (Fig. 1).

2.2. Sweet activity

A series of solutions of mabinlin II varying in concentration were tasted by different people to test its sweetness (Kurihara & Beidler, 1969; Hu & He, 1983; Hu et al., 1985). The lowest concentration inducing taste stimuli was 0.1% or 0.08 mM. The sweetness of 0.8 mM mabinlin II was equivalent to that of 0.3 M sucrose and its sweetness remained undiminished after incubation at 353 K for at least 48 h.

2.3. Crystallization

Crystallization experiments performed using the hanging-drop vapourdiffusion method over a wide range of precipitants and pH values. Two crystal forms were obtained (Fig. 2). Form I crystals were obtained by mixing equal volumes of 5 mg ml⁻¹ protein in 40% ammonium citrate and 0.1 M sodium sulfate in 0.1 M CHES buffer pH 9.8 containing 5% glycerol and equilibrating with the same solution. To obtain form II crystals, 10 mg ml⁻¹ protein solution was mixed with an equal volume of 4.0 M sodium chloride, 5% dioxane in 0.1 M Tris buffer pH 8.5 and was equilibrated with the same solution. Form I grew to approximate dimensions of $0.02 \times 0.2 \times 0.3$ mm after incubation at 277 K for two months and form II grew to dimensions of 0.02 \times 0.02×0.2 mm after incubation at 295 K for more than one month.

2.4. X-ray analysis

Data from form I crystals were collected at room temperature on a MAR 345 imageplate detector mounted on an X-ray generator operating at 40 kV and 50 mA using radiation with a wavelength of 1.5418 Å. Processing of the data from form I crystals using DENZO (Otwinowski & Minor, 1997) showed that the crystals belonged to space group P2, with unit-cell parameters a = 50.16, b = 50.17, c = 76.60 Å, $\beta = 99.6^{\circ}$. The crystal diffracts to 2.8 Å resolution. Assuming four molecules per asymmetric unit, V_m for crystal form I is 1.90 Å³ Da⁻¹, with a corresponding solvent content of 35.3%, which is well within the range normally found in protein crystals (Matthews, 1968). The completeness, R_{merge} and number of reflections with $I > 2\sigma(I)$ in the whole data set were 81%, 0.096 and 7487, respectively. The corresponding values of the completeness and R_{merge} for the last

resolution shell (2.90–2.80 Å) were 72% and 0.16, respectively. Since the form II crystal was too thin for X-ray diffraction using the in-house facility, we hope to study it using synchrotron radiation. Structure determination using the available data for crystal form I and attempts to grow better crystals are in progress.

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References

Caldwell, J. E., Abildgaard, F., Dzakula, Z., Ming, D., Hellekant, G. & Markley, J. L. (1998). Nature Struct. Biol. 5, 427–431.

Ding, M. & Hu, Z. (1986). Acta Bot. Yunnanica, 8, 181–192.

Hahm, Y. T. & Batt, C. A. (1990). Agric. Biol. Chem. 54, 2513–2520.

Hu, Z. & He, M. (1983). Acta Bot. Yunnanica, 5, 207–212.

Hu, Z., Peng, L. P. & He, M. (1985). *Acta Bot. Yunnanica*, **7**, 1–10.

Izawa, H., Ota, M., Kohmura, M. & Ariyoshi, Y. (1996). *Biopolymers*, 39, 95–101.

Kohmura, M. & Ariyoshi, Y. (1998). Biopolymers, 46, 215–223.

Kohmura, M., Nio, N. & Ariyoshi, Y. (1991).
Agric. Biol. Chem. 55, 539–545.

Kondo, K., Miura, Y., Sone, H., Kobayashi, K. & Iijima, H. (1997). *Nature Biotechnol.* 15, 453– 457.

Kurihara, K. & Beidler, L. M. (1969). Nature (London), 222, 1176–1179.

Liu, X. & Hu, Z. (1988). Acta Bot. Yunnanica, 10, 381–388.

Liu, X., Maeda, S., Hu, Z., Aiuchi, T., Nakaya, K. & Kurihara, Y. (1993). Eur. J. Biochem. 211, 281–287.

Matthews, B. W. (1968). J. Mol. Biol. 33, 491–497.Ming, D. & Hellekant, G. (1994). FEBS Lett. 355, 106–108.

Morris, J. A. & Cagan, R. H. (1972). Biochim. Biophys. Acta, 261, 114–122.

Nirasawa, S., Liu, X., Nishino, T. & Kurihara, Y. (1993). *Biochim. Biophys. Acta*, **1202**, 277–280. Nirasawa, S., Masuda, Y., Nakaya, K. & Kurihara,

Y. (1996). *Gene*, **181**, 225–227.

Nirasawa S. Nishino T. Katahira M. Llesuo

Nirasawa, S., Nishino, T., Katahira, M., Uesugi, S., Hu, Z. & Kurihara, Y. (1994). Eur. J. Biochem. 223, 989–995.

Ogata, C. M., Gordon, P. F., de Vos, A. M. & Kim, S.-H. (1992). *J. Mol. Biol.* **228**, 893–908.

Otwinowski, Z. & Minor, W. (1997). Methods Enzymol. 276, 307–326.

Somoza, J. R., Jiang, F., Tong, L., Kang, C.-H., Cho, J. M. & Kim, S.-H. (1993). J. Mol. Biol. 234, 390–404.

Van der Wel, H., Larson, G., Hladik, A., Hladik, C. M., Hellekant, G. & Glaser, D. (1989). Chem. Senses, 14, 75–79.

Van der Wel, H. & Loeve, K. (1972). Eur. J. Biochem. 31, 221–225.

Yamashita, H., Theerasilp, S., Aiuchi, T., Nakaya, K., Nakamura, Y. & Kurihara, Y. (1990). J. Biol. Chem. 265, 15770–15775.